

# Small Angle Neutron Scattering Studies on Polymeric Micelles and Lipids System

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# Outline

- **↓**Why Neutron?
- Instrumentation
- Scattering basics
- Contrast variation
- **+**Time resolved SANS
- **4**Experimental results
- Conclusion

# Why Neutrons?



Non-destructive

Highly penetrating

**4**Wavelength (~1A) in molecular dimensions, so small angle is required to probe large structure

4Energies (~meV) similar to molecular vibrations

↓Isotopically variable scattering power: Deuterium and D<sub>2</sub>O can be used for labelling

# Instrument



**4**Small Angle Neutron Scattering (SANS) probes structure on a scale *d* where

$$d \approx \frac{\lambda}{\theta} \quad (wavelength) \\ (\text{scattering angle}) \\ d = \frac{2\pi}{q} \\ \end{pmatrix}$$

 $\downarrow 0.5 \text{ nm}$   $\downarrow 2.2 \text{ nm}$  (cold noutrons)

George D. Wignall, Physical Properties of Polymers Handbook, Chapter 23, Springer, 2007

# **Scattering basics**

Macroscopic scattering cross section: Normalized by scattering volume

$$\frac{d\Sigma(\vec{q})}{d\Omega} = \frac{N}{V} \frac{d\sigma(\vec{q})}{d\Omega} = \frac{1}{V} \left| \int_{V} \rho(\vec{r}) \exp(i\vec{q}.\vec{r}) d\vec{r} \right|^{2}$$

where  $\rho(r)$  is scattering length density, measure of the interaction of neutron wave with a given nucleus

$$\rho$$
 ( $\vec{r}$ ) =  $\frac{\sum_{j}^{n} b_{j}}{V}$ 

#### **Scattering basics**

Cross section for identical particles:



Structure factor tells us about interactions between particles

**4**Form factor provides information about the structure of individual particles

$$\#S(Q)$$
 is most often found from  $\frac{I_{conc}}{I_{dil}}$ 

**↓**Scattered Intensity: 
$$I(Q) \propto \frac{d L(Q)}{d \Omega}$$

The position and height of peaks can tell us about the system

# **Types of Scattering**





Incoherent scattering:

Can tell us movement of one atom

Is not Q-dependent and contributes only to the noise level, while absorption reduces the overall signal

# **Contrast Variation**







Very sensitive to hydrogenH and D are very different

Scattering lengths:

♣b<sub>H</sub>= -3.74 fermis (coherent), 25.18 fermis (incoherent)

**↓**b<sub>D</sub>= 6.67 fermis (coherent), 3.99 fermis (incoherent)

### **Polymeric micelles**

Polymeric micelles are designed with a branched, hydrophobic interior (core) and hydrophilic exterior (shell) to maintain physical properties characteristic of conventional micelles.

**4**PEP-PEO micelles where PEP constituting the core and PEO constituting the corona (shell)

Star polymer-macromolecules having small core of molecules with branches radiating from the core.

**4**SANS measurements reveal the structural behavior and prove as star polymers



http://en.wikipedia.org/wiki/Micelle http://www.pcc.wur.nl/UK/research/Strategic%20program%20colloids%20and%20interfaces

#### Experimental Results: polymeric micelles I

Features of the star S(Q) w. r. t. peak positions and heights are nearly same for 1<sup>st</sup> and 2<sup>nd</sup> peaks.

**↓**3<sup>rd</sup> peak only for micellar system as for PB star this peak is masked by blob scattering of the polymer arms.

 $43^{rd}$  peak is correlated to the formation of a crystalline phase in micellar system above  $\phi^*$ .

Some increase of micellar S(Q) at low Q indicating some excess scattering

♣ Might arise either from inherent mesoscopic heterogeneities or from weak attractive interactions due to the decreasing solvent quality with increasing molar fraction.



> S(Q) for PEP-PEO micelles with f = 63 (■) compared to the 64 arm PB star (○) at  $\phi/\phi^* \approx 1$ . Solid line: S(Q) from theory

## **Experimental Results : polymeric micelles II**

 $\clubsuit$  Good agreement between theory and expt. for liquid-solid boundary for all f and  $\eta$ 

Liquid to bcc crystal transition is perfectly reproduced experimentally for f<70.

4Packing fraction for crystallization at η≈0.21



#### **Experimental Results : polymeric micelles III**

For the first peak of S(Q) crossing  $\phi^*$  with f=67 is bigger than 2.8, minimum value for a freezing transition according to the Hansen-Verlet criterion.

**H**This implies the disordered phase as a glass.

**4**The peak is further growing with increasing  $\phi$ .

 $\clubsuit$  for  $\phi > \phi^*$  a third small peak is forming, and the position ratio between the three peaks is corresponding to a simple cubic (sc) or bcc lattice.

+Inset: 2-dimensional SANS detector picture of f = 63,  $\phi = 0.16$  evidencing Bragg reflections.



M. Laurati et al., Physical Review E 76, 041503, 2007

# Lipid bilayers: excellent for cell membrane

- Composed of lipid molecules: 4-5nm thick.
- **4** Lipid molecules spontaneously form bilayer when hydrate
- **Phospholipid**: A hydrophilic polar head group Two Hydrophobic acyl chains
- Hydrophobic interaction is the driving force
- Tendency to close on themselves
- **4** Extensive; upto millimetres









Avanti Polar Lipids (DMPC)

http://www.avantilipids.com/ProductStructures.asp?n=850345 http://www.rpi.edu/dept/bcbp/molbiochem/MBWeb/mb1/part2/lipid.htm Describe as sending the neutron beam into a sample and watching how it is temporarily broaden when reaching the detector after passing through the sample

**4**Useful for kinetics studies because perturbations are inevitable in most experimental techniques

**4** If time scale is shorter - time slicing required

**↓**With chopper at the source it is possible to get 50µs - 100ms time resolution.

## **Experimental Result: Lipid kinetics I**

#Principle of the detection of lipid kinetics by TR-SANS

Here two LUV's, D/H-LUV exhibited little scattering, D/H-LUV is "invisible" for neutrons in this contrast-matching condition [Fig. (b)].

**4**Fig. (c) shows how the lipid exchange between D- and H-LUV reduces the difference in the SLD of LUV's from solvent (i.e., contrast) with time, this leads to a decrease in the scattering intensity.



#### **Experimental Results: Lipid kinetics II**

The scattering from the mixture of D-LUV and H-LUV faded over time indicating lipid exchange in the observable time scale.

SANS profile shows identical shape for all times indicating the shell construction of LUV is maintained during lipid exchange process.



## **Experimental Results : Lipid kinetics III**

**4**Contrast decays of LUV's after mixing Dand H-LUV at four different temperatures.

**4**The contrast decays more sharply with the increase of temperature.

Inset shows the contrast decay over a longer period at 27.1°C.



Minora Nakano et al., Phys. Rev. Lett. 98, 238101 (2007)

## **Experimental Results : Lipid kinetics IV**

The half-lives ( $t_{1/2} = (\ln 2/k)$ ) of the lipid exchange at 50 and 30°C were estimated at 0.67 and 5.4 h, respectively, which are close to or slightly smaller than those obtained using a radioisotope (0.74 and 9.6 h, respectively).

Arrhenius plots of the obtained parameters exhibited a good linear relationship.



Arrhenius plots of the rates of intervesicular exchange  $(k_{ex})$  and flip-flop  $(k_f)$ .

Minora Nakano et al., Phys. Rev. Lett. 98, 238101 (2007)

# **Drawbacks of SANS**

- SANS is a routine technique available at neutron-scattering facilities associated with research nuclear reactors.
- No home version of this technique.
- Neutron sources are very expensive to build and to maintain.
- Another problem with this technique is that neutron flux is very low.
- The interaction of neutrons with matter is weak.

#### Conclusion

- Small-angle neutron scattering (SANS) is a very popular method used by physicists, material scientists, chemists and biologists.
- SANS can determine structures, phase transitions, and morphology.
- Possible to do experiments in bulk systems
- TR-SANS is a new method to determine kinetics studies of lipid and polymeric micelles precisely.

Thank you Questions?